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13.2 The Nature of Liquids The general properties of solids reflect the orderly arrangement of their particles and the fixed locations of their particles In most solids, the atoms, ions, or molecules are packed tightly together. When you heat a solid, its particles vibrate more rapidly as their kinetic energy increases.

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Activities for chapter 13: States of matter •What do I already know about states of matter?(index card) and Vocabulary table •Chapter 13 reading guide (feb break assignment) and Powerpoints
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The kinetic molecular theory Pressure The Boltzmann distribution

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