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If solid settles to the
bottom filter the

solution and mass the
remaining solid, if the

mass is the same then

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the solution was

saturated, if the mass is lower than it is an unsaturated solution.

Model 2 - Solute

Dissolved vs. Solute

Added The following

data refer to an

experiment in which a

measured mass of solid

is added to 10.0 g of

20 °C water.

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amount of solute

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following variables in

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the experiment in
Model 2 Dependent
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your answer, consider
the external and
internal pressure data
given in Model 1. 2

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Solubility - Chemistry

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solution is a solution
that contains less than
the maximum amount
of solute that is
capable of being
dissolved.

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When the solution

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equilibrium point is reached and no more solute will dissolve, the solution is said to be saturated. A saturated solution is a solution that contains the maximum amount of solute that is capable of being dissolved. At 20°C , the maximum amount of NaCl that will dissolve in 100. g of water is 36.0 g.

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3 6. Have each person
in your group provide
an example of the
word “saturated” as it
is used in an everyday
context. Summarize
the meaning of the
word in the space

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below. 7. Use a

grammatically correct sentence to explain why beakers D and E in Model 1 are labeled as “saturated.”

26 Saturated and Unsaturated Solutions-S

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Saturated fats and unsaturated fats are found in a variety of foods. The type of fat

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Saturated And
you consume,

especially if you're
trying to lower the
amounts of lipids in

Solutions
Answers
your diet, can be
confusing. The

American Heart
Association (AHA)

recommends that

between 20% and 35%
of your total daily ...

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