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Use the periodic table to identify each  
element described below. 1.

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the AtomThe Structure of the Atom  
CHAPTER 4 SOLUTIONS MANUAL Section  
4.1 Early Ideas About Matter pages  
102–105 Section 4.1 Assessment page  
105 1. Contrast the methods used by the  
Greek philo-sophers and Dalton to study  
the atom. Greek philosophers could not  
conduct

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naturally occurring magnesium is as  
follows: 79% magnesium-24, 10%  
magnesium-25, and 11%  
magnesium-26. Calculate the atomic

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4 Cl(s). How many grams of NH<sub>3</sub> (g)

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## Answers

reacted? Is the law of conservation of mass observed in the above reaction? Justify your answer. 157.5 g 106.5 g 51 g  
Yes. Mass of reactants equals mass of products. Section 3.2 Assessment page 79 10. Classify each example as a physical change or a chemical change.  
a. crushing an aluminum can physical

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## Answers

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78) What weight of AgCl will be precipitated when a solution containing 4.77 g of NaCl is added to a solution of 5.77 g of AgNO<sub>3</sub>?

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