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In chemistry, a solution is a special type of homogeneous mixture which is composed of 2 or more substances. The term aqueous solution is used for the solutions where one of the solvents is water. In such a mixture, a solute is a substance dissolved in another substance, known as a solvent. Chemistry Solutions MCQs. 1.

Chemistry Solutions MCQs

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SSC Chemistry MCQ Question With Answer 2021

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Chemistry MCQs for Class 12 with Answers Chapter 2 Solutions

25.3 g of sodium carbonate, Na 2 CO 3 is dissolved in enough water to make 250 ml of solution. If sodium carbonate dissociates completely, molar concentration of sodium Ion, Na + and carbonate ions, CO 3 2 - are respectively (Molar mass of Na s CO 3 = 106 g mol - 1) [CBSE AIPMT 2010]

Solutions: MCQs | Question - Paper 2

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